

Para-Substituted Functionalised Ferrocene Esters with Novel Antibacterial Properties

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ABSTRACT

Introduction: Bacterial antibiotic resistance is on rise despite advances in the development of new antibiotics. In an attempt to circumvent resistance, scientists are shifting focus from modifying existent antibiotics to identifying new antibiotic compounds.

Aim: To assess the potential antibiotic effects of functionalised ferrocenecarboxylates para-substituted on the phenoxy pendant group to form: 4-fluorophenyl, 4-chlorophenyl, 4-bromophenyl, 4-iodophenyl and 4-(H-pyrrol-1-yl)phenyl.

Materials and Methods: For this, we employed the Kirby-Bauer disc diffusion method using a collection of nine bacterial species: *Staphylococcus aureus*, *Escherichia coli*, *Micrococcus luteus*, *Pseudomonas aeruginosa*, *Serratia marcescens*, *Klebsiella*

pneumoniae, *Bacillus subtilis*, *Proteus vulgaris* and *Enterobacter aerogenes*.

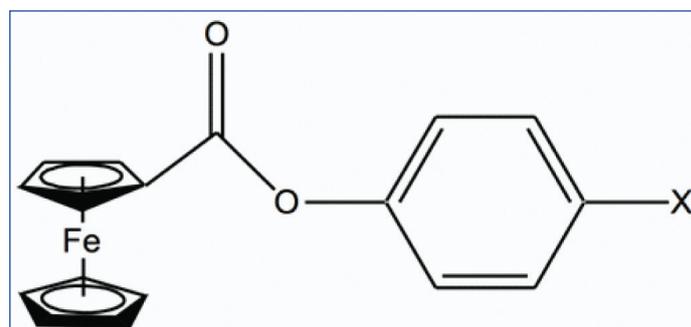
Results: The results show that all four-halogen substituted ferrocenecarboxylates 4-fluorophenyl (23.33 μ M, 11.66 μ M, 5.83 μ M), 4-chlorophenyl (10.16 μ M, 5.08 μ M, 2.54 μ M), 4-bromophenyl (9.0 μ M, 4.5 μ M, 2.25 μ M), and 4-iodophenyl (17.12 μ M, 8.56 μ M, 4.28 μ M) exhibited an antibacterial effect by reducing proliferation of *Bacillus subtilis*. Meanwhile, only 4-bromophenyl (9.0 μ M) and 4-chlorophenyl (10.16 μ M) ferrocenecarboxylates were able to decrease the growth of *Micrococcus luteus*.

Conclusion: Hence, functionalised ferrocenecarboxylates para-substituted with small and simple groups represent a novel class of bio-organometallic compounds with the potential to be used as antibacterial agents.

Keywords: Antibacterial, Bio-organometallic compounds, Phenoxy pendant, Ferrocenecarboxylate

INTRODUCTION

The over use of antibiotics and accelerated microevolution of bacteria have created a major threat to healthcare. Bacteria have developed antibiotic resistance genes through random mutations and have capitalised on horizontal gene transfer to facilitate the creation of bacteria strains with de novo antibiotic resistance [1]. Resistance to one antibiotic has even been shown to influence how a bacterium responds to another non related antibiotic [2]. Furthermore, it has been shown that Gram positive bacteria have developed resistance to antibiotics prior to their commercialisation via non anthropological exposure [3]. Bio-organometallic chemistry is developing a myriad of alternatives with compounds exhibiting unprecedented modes of actions and potential of combating antibiotic-resistant bacterial strains. It has been shown that the modification of organometallics with existing antibiotics may be able to overcome antibiotic resistance [4]. For instance, organometallic structures such as functionalised ferrocenes, exhibit a combination of biocompatibility, lipophilicity and reduction-oxidation (redox) properties that hold great promise in biomedicine. These structures present a platform for scaffolding and functionalisation of fathomless possibilities, some of which have already been shown to be successful novel antitumour, anti-parasitic, antifungal, and antimalarial alternatives [5-10]. For example, ferroquine, a ferrocene derivative of the drug chloroquine, has shown to overcome the drug resistance developed by malarial parasites to common antimalarial drugs [11]. However, despite the aforementioned advances in bio-organometallic research, studies evaluating the antibacterial activity of ferrocenecarboxylates are scarce. This study focused on the antibacterial effect of the following functionalised ferrocenes para-substituent on the phenoxy pendant group [Table/Fig-1]: 4-fluorophenyl-ferrocenecarboxylate (Fc-4F-Ph), 4-chlorophenyl-ferrocenecarboxylate (Fc-4Cl-Ph), 4-bromophenyl-ferrocenecarboxylate (Fc-4Br-Ph), 4-iodophenyl-



[Table/Fig-1]: Structure of ferrocenecarboxylate esters. "X" of the phenoxy pendant group indicates substitution site X=F, Cl, Br, I, or Py.

ferrocenecarboxylate (Fc-4I-Ph), and their analog replacing the halide with a redox active species 4-(1H-pyrrol-1-yl)phenyl ferrocenecarboxylate (Fc-Py). The Kirby-Bauer disc diffusion susceptibility method was used to assess possible antibacterial activity of functionalised ferrocenes on *Staphylococcus aureus*, *Escherichia coli*, *Micrococcus luteus*, *Pseudomonas aeruginosa*, *Serratia marcescens*, *Klebsiella pneumoniae*, *Bacillus subtilis*, *Proteus vulgaris* and *Enterobacter aerogenes*.

This study represents the first of its kind, reporting the antibacterial properties of para-substituted ferrocenecarboxylates using a diverse gamut of microorganisms. The studied organisms encompass both Gram-positive and Gram-negative bacteria from various genera, of varied morphologies and metabolisms. These compounds may represent alternatives for treatment of antibiotic resistant strains and possibly reduce their burden on public health systems.

MATERIALS AND METHODS

The present in vitro study was conducted from August 2015 to May 2016 at the Biosafety Level 2 PRISE Research laboratory, Department of Biology at the University of Puerto Rico in Ponce.

Bacterial Cultures

Nine bacterial species were tested. These were inoculated on solid growth media and cultured in 100 mm petri dishes under their respective conditions and nutritional requirements [Table/Fig-2]. These organisms were obtained from ATCC commercial strains, previously maintained and characterised by the Department of Biology, University of Puerto Rico, Ponce.

Organism	Agar Media	Temperature (°C)	Incubation Time (hours)	Antibiotic Concentration
<i>S. aureus</i> ATCC® 6538™	Mannitol salt	37°0	24	Penicillin (500 I.U./mL)
<i>E. coli</i> ATCC® 25822™	Luria Bertani	37°0	24	Kanamycin (50 µg/mL)
<i>M. luteus</i> ATCC® 10240™	Nutrient	30°0	72	Penicillin (500 I.U./mL)
<i>P. aeruginosa</i> ATCC® 27853™	<i>Pseudomonas</i>	37°0	24	Kanamycin (50 µg/mL)
<i>S. marcescens</i> ATCC® 14756™	Nutrient	25°0	48	Kanamycin (50 µg/mL)
<i>K. pneumoniae</i> ATCC® 13883™	MacConkey	37°0	24	Kanamycin (50 µg/mL)
<i>E. aerogenes</i> ATCC® 13048™	Nutrient	30°0	24	Kanamycin (50 µg/mL)
<i>P. vulgaris</i> ATCC® 8427™	MacConkey	37°0	48	Kanamycin (50 µg/mL)
<i>B. subtilis</i> ATCC® 6633™	Nutrient	30°0	24	Kanamycin (50 µg/mL)

[Table/Fig-2]: The culturing conditions and antibiotic control used in study for each bacterium. *Staphylococcus aureus*, *Escherichia coli*, *Micrococcus luteus*, *Pseudomonas aeruginosa*, *Serratia marcescens*, *Klebsiella pneumoniae*, *Enterobacter aerogenes*, *Proteus vulgaris* and *Bacillus subtilis*.

Synthesis of Ferrocene Esters

The ferrocene esters used in this study 4-fluorophenyl, 4-chlorophenyl, 4-bromo, 4-iodo, and 4-(1H-pyrrol-1-yl)-phenyl were synthesised and characterised as previously detailed [11]. In summary, ferrocenecarboxylic acid was dissolved in dry dichloromethane at room temperature under a nitrogen atmosphere. To this solution, oxalyl chloride was added drop wise and stirred overnight and the resulting solution was filtered. Separately, under a nitrogen atmosphere the para-substituted phenol vector was dissolved in dichloromethane; the ferrocenecarbonyl chloride filtrate previously prepared, was added drop wise to this solution while it stirred over the course of 6-12 hours. The reaction was monitored using Thin Layer Chromatography (TLC) and the mixture was filtered through a celite pad. The filtrate was washed with 0.1M HCl and purified by column chromatography using silica as the stationary phase and dichloromethane as the mobile phase. Characterisation was performed by Nuclear Magnetic Resonance (NMR) spectroscopy, Infrared (IR) spectroscopy, and elementary analysis.

Ferrocenecarboxylates Treatments

The six ferrocenes compounds tested; Fc-4Br-Ph, Fc-4Cl-Ph, Fc-4F-Ph, Fc-4I-Ph, Fc-4Py-Ph, and Fc-Ph were prepared by suspending a total of 1.3 mg of each compound in 150 µL of Dimethyl Sulfoxide (DMSO) (vehicle) and then agitated at 1,200 rpm for one hour at room temperature. After which the compounds were diluted with sterile cell culture grade water to their final concentrations listed in [Table/Fig-3].

Kirby-Bauer Disc Diffusion

Compounds were first screened to optimise volume and conditions with one positive control and one disc per compound concentration. For experiments each petri plate was divided into five sections including one as a negative vehicle control (DMSO+cell culture

Ferrocenes Compounds	Concentration I µM	Concentration II µM	Concentration III µM
Fc-4F-Ph	23.33	11.66	5.83
Fc-4Br-Ph	9.00	4.50	2.25
Fc-4I-Ph	17.12	8.56	4.28
Fc-4Cl-Ph	10.16	5.08	2.54
Fc-4Py-Ph	9.04	4.52	2.26
Fc-Ph	24.80	12.40	6.20

[Table/Fig-3]: Ferrocene compounds with the concentrations used in this study expressed in µM.

grade water), one with an antibiotic for positive inhibition control (varies by organism) and three for each concentration (I, II, and III) of the six ferrocenes compounds tested [Table/Fig-4]. A total of 50 µL of broth media containing active bacteria cultures were used to inoculate each plate. Thirty minutes after inoculation, sterile filter paper discs with a diameter of 0.63 mm were placed in triplicates for each compound concentration in their appropriate sections and a single disc was placed in the control sections. This was followed by the addition 10 µL of each corresponding ferrocene or control solution directly to the center of the filter disc. The plates were incubated under the conditions specified in [Table/Fig-2], the plates were evaluated for bacterial growth inhibition. The inhibitory effect was determined by comparing the presence or absence of a halo formed around the discs to that of the negative and positive controls with those of the experimental treatments [Table/Fig-2].

Bacteria	Concentration I	Concentration II	Concentration III
<i>B. subtilis</i>			
Fc-4Br-Ph	44.5±1.9	41.6±1.3	43.8±1.5
Fc-4Cl-Ph	43.1±0.6	45±1.2	42.6±1.1
Fc-4I-Ph	46.6±2.2	43.3±1.1	45.5±1.5
Fc-4F-Ph	48.3±3.2	-	-
<i>M. luteus</i>			
Fc-Br	42.48±1.4	-	-
Fc-Cl	44.68±1.2	-	-

[Table/Fig-4]: Effective percentage of inhibition of *B. subtilis* and *M. luteus* by ferrocenes concentrations relative to positive controls.

Data shown in percentage ± standard error of the mean for each concentration. (-): no inhibition observed

STATISTICAL ANALYSIS

Descriptive statistical analysis was performed using IBM SPSS Statistics SPSS Version 21.0. (IBM Corp., NY, USA). Experiments were conducted in technical triplicates and with biological duplicates. When present, the diameter (mm) of the inhibition halo was measured and the values of the triplicates were averaged. The inhibitory effects of the ferrocenecarboxylates were calculated by dividing the average diameter of the halos of each compound by that of the control antibiotic and expressed as a percentage±standard error of the mean.

RESULTS

The growth of *B. subtilis* was inhibited by all three concentrations of Fc-4Cl-Ph (10.16 µM, 5.08 µM, 2.54 µM); Fc-4Br-Ph (9.0 µM, 4.5 µM, 2.25 µM), Fc-4I-Ph (17.12 µM, 8.56 µM, 4.28 µM), and the highest concentration of Fc-4F-Ph (23.33 µM). Growth of *M. luteus* was inhibited by Fc-4Br-Ph (9.0 µM) and Fc-4Cl-Ph (10.16 µM) [Table/Fig-4]. Outcomes of treating nine bacterial isolates with ferrocenecarboxylates are listed in [Table/Fig-5].

DISCUSSION

The present microbiological study, determined the ability of ferrocenecarboxylates to inhibit bacterial growth by quantifying the halo of inhibition, measuring its diameter in mm. The observed antibacterial activity towards most of the tested Gram-positive

Bacterium	Gram Reaction	Positive Control	Fc-Br	Fc-Cl	Fc-F	Fc-I	Fc-Ph	Fc-Py
<i>S. aureus</i>	+	Penicillin	-	-	-	-	-	-
<i>E. coli</i>	-	Kanamycin	-	-	-	-	-	-
<i>M. luteus</i>	+	Penicillin	+ I	+ I	-	-	-	-
<i>P. aeruginosa</i>	-	Kanamycin	-	I, II, III NP	-	I, II, III NP	-	-
<i>S. marcescens</i>	-	Kanamycin	-	-	-	-	N/A	-
<i>K. pneumoniae</i>	-	Kanamycin	-	-	-	-	N/A	-
<i>B. subtilis</i>	+	Kanamycin	+ I, II, III	+ I, II, III	+ I	+ I, II, III	N/A	-
<i>P. vulgaris</i>	-	Kanamycin	-	-	-	-	N/A	-
<i>E. aerogenes</i>	-	Kanamycin	-	-	-	-	N/A	-

[Table/Fig-5]: Summary of organisms used and the observed results of inhibition by compound.

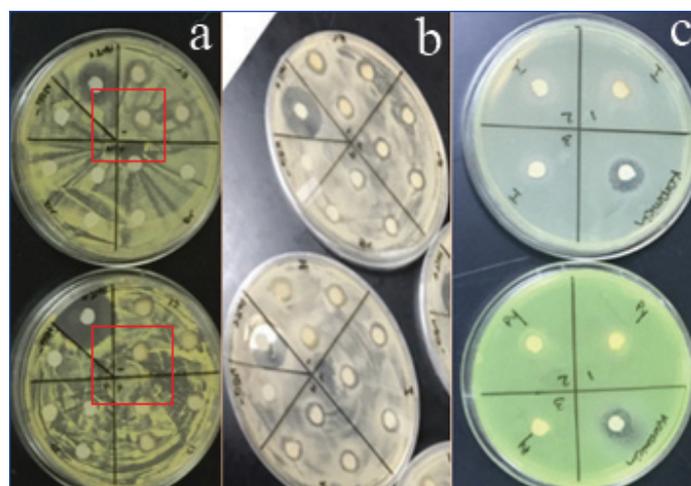
4-fluorophenyl ferrocenecarboxylate (Fc-4F-Ph), 4-chlorophenyl ferrocenecarboxylate (Fc-4Cl-Ph), 4-bromophenyl ferrocenecarboxylate (Fc-4Br-Ph), 4-iodophenyl ferrocenecarboxylate (Fc-4I-Ph), and their analog replacing the halide with a redox active specie 4-(1H-pyrrol-1-yl)-phenylferrocenecarboxylate (Fc-Py). Bacteria, gram-staining, antibiotic controls and concentrations (I: concentration I; II: concentration II; III: concentration III) used of ferrocenecarboxylates, with corresponding outcome.

(-): No inhibition observed; (+): Inhibition observed; NP: No pigmentation; N/A: not assayed

bacteria can be attributed mainly to the structural differences between Gram-positive and Gram-negative bacteria in their cytoplasmic lipid membrane composition. Gram-positive bacterium possesses a thicker membrane with a diverse lipid profile that may increase ferrocene-membrane interactions [12]. Ferrocene compounds are lipophilic and allowing better penetration across the more lipophilic membrane present in the Gram-positive bacteria [13]. The organometallic redox active species Fc-Py did not showed any antibacterial activity against microorganisms tested, only ferrocene complexes with halogenated phenols demonstrated antibacterial properties [Table/Fig-4]. For example, fluorine has been shown to increase the biological effect of an organic compound. N-(2-chloro-5-nitrobenzylidene)-2-fluorobenzohydrazide which showed an effective antibacterial activity (against *Bacillus subtilis*, *Staphylococcus aureus*, *Escherichia coli*, and *Pseudomonas fluorescens*) attributed to the electronegativity of the fluorine [14]. A patent submitted showed new nitroimidazole derivatives with iodophenol to treat mycobacterial infections [15]. Bromophenols extract from marine algae showed growth inhibition of *Botrytis cinerea*, *Pseudomonas fluorescens*, *Staphylococcus aureus*, *Vibrio cholera*, and *Proteus mirabilis* [16]. 2-benzyl-4-chlorophenol is a well-known germicidal that recently was isolated as a natural product from *Shewanella halifaxensis* IRL548 [17]. All these active organic compounds have halogen atoms substituted on to the phenol side group, with the of exception of Fc-Py.

Previous studies, have demonstrated antibacterial, antimalarial, and antifungal properties of ferrocenes derivatised with large complex groups or substituted with known antimicrobials [4, 13-15, 17]. On the contrary, these compounds do not contain such substitutions, which may be susceptible to degradation and inhibition by microorganisms. It has been demonstrated that ferrocenoyl 17 β -hydroxy-estra-1,3,5-(10)-trien-3-olate (ferrocene ester containing estradiol pendant group) exhibits anti-proliferative activity of with an IC₅₀ ranging from 1.4-256 μ M in the MCF-7 breast cancer cells and 1.8-500 μ M in the MCF-10A non-malignant cell line [11]. These organometallic complexes demonstrated compatibility with biological systems; for this reason, this study explored bacterial growth, inhibition properties.

Although there was no observed inhibition of *P. aeruginosa* growth, Fe-4I-Ph and Fe-4Cl-Ph inhibited its pigmentation at all three tested concentrations [Table/Fig-6]. While it is unclear how these compounds interfered with the pigmentation of *P. aeruginosa*, mutations in pigmentation genes such as pyocyanin, have been linked to antibiotic resistance and resistance to phages [18]. Given



[Table/Fig-6]: Representative images of kirby-bauer test. Panel A) *Micrococcus luteus*; positive results indicated in red Fe-Br (9.00 μ M) (top) and Fe-Cl (10.16 μ M) (bottom) Panel B) *Bacillus subtilis*; positive results at all concentrations of Fe-Br (9.00 μ M, 4.50 μ M, and 2.25 μ M) and Fe-I (17.12 μ M, 8.56 μ M, and 4.58 μ M) Panel C) *Pseudomonas aeruginosa*; Fe-I interfering with the *P. aeruginosa* (top) and expected pigmentation (bottom), although there was no bacterial inhibition observed. *indicates positive antibiotic control and ** indicates vehicle control.

the apparent importance of *P. aeruginosa* pigmentation the inhibition observed should be further studied.

LIMITATION

This study demonstrates the proof of concept use of novel ferrocenecarboxylates as antibiotics for nine bacteria. Further studies should be conducted on other pathogenic organisms such as fungi to address plausible clinical applications. In addition, there may be differences in diffusion rate between these ferrocenecarboxylate compounds that may be considered in future studies specially in newly emerging resistant bacteria.

CONCLUSION

This study demonstrates that the anti-proliferative effects of 4-bromophenyl ferrocenecarboxylates (Fe-Br) are not exclusive to eukaryotes. Further studies should be conducted into the bioactive properties of these functionalised ferrocenecarboxylates to decipher pathways and modes of action. In addition, the observed effects were not broad spectrum across similar bacteria as were penicillin and kanamycin, indicating an underlying specificity. Where these compounds bind to and how this interaction interfere with growth has yet to be elucidated and could be the subject of future studies.

ACKNOWLEDGEMENTS

The bacteria used in this study were provided by the Department of Biology of the University of Puerto Rico in Ponce. This work was supported by the UPR-PRISE Program NIH-NIGMS Grant #R25GM096955.

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Date of Submission: **May 24, 2017**
Date of Peer Review: **Jun 21, 2017**
Date of Acceptance: **Sep 29, 2017**
Date of Publishing: **Feb 01, 2018**

FINANCIAL OR OTHER COMPETING INTERESTS: None.